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Reactions of Cp*MCl(P-O) (M = Rh, Ir) bearing P-O coordination with ethynylbenzene derivatives having an electron-withdrawing group in the presence of a PF₆ anion result in either insertion into a P-C bond or a transannular addition between a metal and an ipso-carbon of the phosphine.

When transition metal halide complexes are allowed to react with alkynes in the presence of anions such as PF₆, BF₄, etc., it is well-known that vinylidene or carbene complexes may be formed, depending on the solvent. However, we have recently reported that the unprecedented insertion of alkyne into a metal-oxygen σ-bond can occur without the formation of vinylidene complexes.² On treatment of the rhodium complex [Cp*RhCl(MDMPP- $\kappa^2 P$,O)] **1a** (MDMPP- $\kappa^2 P$,O = κ O-1-O- κP -2-PPh₂-3-MeOC₆H₃) with mono- and di-substituted alkynes (HC≡CPh, HC≡CCOOR or ROOCC≡CCOOR; R = Me, Et) in the presence of the PF₆⁻ anion, single and double insertion of alkynes into the Rh-O bond occurred to form various types of complex, depending on the alkyne, as follows; (1) a complex bearing five- and six-membered rings arising from a double insertion, (2) a seven-membered metallacycle bearing a CO ligand arising from a single insertion and an extraction of CO from an ester and (3) a neutral complex bearing a sevenmembered ring arising from a single insertion.²

When ethynylbenzene derivatives bearing an electronwithdrawing substituent at the 4-position of the phenyl group

† Electronic supplementary information (ESI) available: spectroscopic data for complexes 3-5. See http://www.rsc.org/suppdata/dt/b0/ b004984k/

were treated with [Cp*MCl(MDMPP- $\kappa^2 P, O$)] (1a: M = Rh;³ **1b**: $M = Ir^4$) or $[Cp^*RhCl(BDMPP-\kappa^2P, O)]^2$ **2a**⁵ (BDMPP- $\kappa^2 P, O = \kappa O - 1 - O - \kappa P - 2 - PRPh - 3 - MeOC_6H_3$ bearing a P, Obidentate ligand derived from (2,6-dimethoxyphenyl)diphenylphosphine (MDMPP) and bis(2,6-dimethoxyphenyl)phenylphosphine (BDMPP) in the presence of KPF₆, we found that novel types of reaction: an insertion into a P-C bond or a transannular addition between a Rh atom and an ipso-carbon atom of the phosphine ligand, occurred readily. We report here the reactions with ethynylbenzene derivatives bearing an electronwithdrawing substituent such as the COOMe or NO₂ group.

Reaction of 1a with HC≡CC₆H₄COOMe-4 at room temperature in the presence of KPF₆ in acetone-CH₂Cl₂ gave two compounds by crystallization from CH₂Cl₂-diethyl ether: 3a ‡ and 4a‡ as confirmed from elemental analyses and FAB mass spectroscopy (Scheme 1). X-Ray analysis revealed that 3a contained a (P,O,C) tridentate ligand resulting from the head-tohead double-insertion of alkyne into the Rh-O bond.§ Two carbon atoms bearing the phenyl group were connected to the Rh and O atoms. The infrared spectrum showed bands due to methoxycarbonyl and PF₆ groups at 1717 and 839 cm⁻¹, respectively. The ¹H NMR spectrum in CD₂Cl₂ showed one doublet at δ 1.36 due to the Cp^* protons and three singlets at δ 3.35, 3.88 and 3.95; the former is due to the methoxy group and the others are due to the methoxycarbonyl groups. The $^{31}P\{^{1}H\}$ NMR spectrum showed a doublet at δ 35.84 $(J_{\rm RhP} = 157.8 \text{ Hz}).$

The infrared spectrum of 4a showed three characteristic bands at 3285, 1717 and 1605 cm⁻¹ due to the OH, carbonvl groups and C-C double bond, respectively. In the ¹H NMR spectrum a broad resonance due to a hydroxyl proton, which

Scheme 1 Reactions of 1 and 2 with $HC \equiv CC_6H_4R'-4$ ($R = 2,6-(MeO),C_6H_3$, R' = COOMe or NO_2).

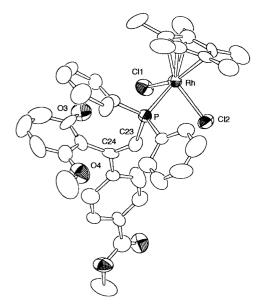


Fig. 1 Molecular structure of 3a. Selected bond lengths (Å) and angles (°): Rh(1)-P(1) 2.341(5), Rh(1)-Cl(1) 2.428(5), Rh(1)-Cl(2) 2.411(5); P(1)–C(23) 1.83(2), C(23)–C(24) 1.32(2); P(1)–Rh(1)–Cl(1) 87.1(2), P(1)-Rh(1)-Cl(2) 85.0(2), Cl(1)-Rh(1)-Cl(2) 93.4(2), P(1)-C(23)-C(24) 135(1), C(23)-C(24)-C(25) 117(1).

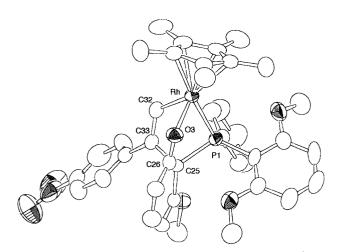


Fig. 2 Molecular structure of 5ad. Selected bond lengths (Å) and angles (°): Rh(1)-P(1) 2.304(2), Rh(1)-O(3) 2.106(4), Rh(1)-C(32) 2.013(6), O(3)–C(26) 1.263(7), C(25)–C(26) 1.505(8), P(1)–C(25) 1.943(6), C(25)–C(33) 1.567(8), C(32)–C(33) 1.340(8); P(1)–Rh(1)–O(3) 81.1(1), P(1)-Rh(1)-C(32) 76.3(2), O(3)-Rh(1)-C(32) 85.2(2), Rh(1)-P(1)–C(25) 90.2(2), P(1)–C(25)–C(26) 105.1(4), C(25)–C(26)–O(3) 118.7(5), Rh(1)–O(3)–C(26) 117.9(4), Rh(1)–C(32)–C(33) 120.4(4), C(32)-C(33)-C(25) 114.2(5), P(1)-C(25)-C(33) 96.4(4).

disappeared on treatment with D_2O , appeared at δ 6.74 in addition to three characteristic methyl resonances at δ 1.41(d), 3.44(s) and 3.83(s), due to 1,2,3,4,5-pentamethylcyclopentadienyl, methoxy and methoxycarbonyl groups, respectively. There were no signs of a PF₆ group in the infrared and ³¹P{¹H} NMR spectra, suggesting that the complex was neutral. The structure was confirmed by X-ray analysis (Fig. 1).¶ As expected, the molecule is neutral and the rhodium atom is surrounded by two chlorine atoms and one phosphorus atom. Cis-insertion of alkyne into the P-C bond has occurred, accompanied by cleavage of a Rh-O bond. The Rh-P bond length of 2.341(5) Å is longer than that of **1a** by 0.04 Å, due to the absence of chelation and is 0.02 Å shorter than that (2.366(1) Å) of Cp*RhCl₂(MDMPP), probably due to lower steric demand than in the MDMPP ligand.³ A similar reaction occurred on treatment with 1b in acetone-CH2Cl2 to give 3b and **4b** (Scheme 1).‡ The fact that a similar reaction in MeOH gave exclusively 3a as the only isolated complex confirmed that the chlorine atom in 4 originated from dichloromethane.

Complex 2a containing the bulky phosphine BDMPP was treated with $HC = CC_6H_4R-4$, (R = COOMe or NO_2) in acetone-CH₂Cl₂ in the presence of KPF₆ at room temperature (Scheme 1) to yield complexes 5ac (yellow, R = COOMe) and **5ad** (brown, $R = NO_2$) as confirmed from elemental analyses and FAB mass spectroscopy.‡ X-Ray analysis of 5ad revealed that the structure consists of five- and six-membered rings derived from a transannular insertion of alkyne between the Rh atom and the ipso-carbon atom of the phosphine ligand, accompanying the subsequent transformation of the Rh-O σ -bond to a Rh←O coordination (Fig. 2).|| The change of this bonding mode caused an elongation of ca. 0.06 Å in the Rh–O bond length.

The IR spectrum showed bands due to methoxycarbonyl and ketone groups at 1715 and 1630 cm⁻¹ for 5ac and due to a ketone group at 1628 cm⁻¹ for **5ad**, respectively. The ¹H NMR spectrum showed three methoxy groups at δ 2.97 (s), 3.06 (bs) and 3.49 (bs) and one methoxycarbonyl group at δ 3.84 (s) for **5ac** and three methoxy groups at δ 2.99 (s), 3.08 (bs) and 3.52 (bs) for 5ad, respectively. A remarkable feature is that the ³¹P{¹H} NMR doublets show large downfield shifts (ca. δ 140).

Further mechanistic studies are now in progress.

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Notes and references

‡ Elemental analyses of all new complexes prepared here are in good agreement with the calculated values. Elementary analyses and spectroscopic data are enclosed in the electronic supplementary information.† § Crystal data for **3a**: C₄₉H₄₇O₆P₂F₆Rh, M = 1010.8, monoclinic, space group $P2_1/a$ (no. 14), a = 15.534(3) Å, b = 19.986(3) Å, c = 15.757(3) Å, β = 103.70(1)°, V = 4752(1) ų, Z = 4, D = 1.413 g cm⁻³, μ = 4.96 cm⁻¹ (MoKα), F(000) = 2072, T = 298 K. Data were collected on a Rigaku AFC5S difractometer. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on F^2 to give R1 = 0.064 for 3298 reflections and $R_{\rm w} = 0.198$ for 8328 reflections.

¶ Crystal data for 4a: $C_{39}H_{42}O_6PCl_2Rh$, M = 811.5, orthorhombic, space group $Pca2_1$ (no. 29), a = 34.835(7) Å, b = 9.135(8) Å, c = 12.305(5) Å, V = 3915(5) Å, Z = 4, D = 1.377 g cm⁻³, $\mu = 6.55$ cm⁻¹ (MoK α), F(000) = 1672, T = 300 K. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on F^2 to give R1 = 0.068 for 2216 reflections and $R_{\rm w} = 0.203$ for 3607 reflections.

|| Crystal data for **5ad**: $C_{39}H_{46}NO_9P_2F_6Rh$, M = 951.6, monoclinic, space group $P2_1/a$ (no. 14), a = 15.299(7) Å, b = 14.826(7) Å, c = 19.505(9) Å, $\beta = 106.78(3)^\circ$, V = 4235(3) Å³, Z = 4, D = 1.492 g cm⁻³, $\mu = 5.57 \text{ cm}^{-1} \text{ (MoK}\alpha), F(000) = 1952, T = 298 \text{ K. The structure was}$ solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on F^2 to give R1 = 0.065 for 4458 reflections and $R_{\rm w} = 0.180$ for 7708 reflections. CCDC reference number 186/2117. See http://www.rsc.org/suppdata/ dt/b0/b004984k/ for crystallographic files in .cif format.

- 1 (a) M. I. Bruce, Chem. Rev., 1991, 91, 197 and refs. therein; (b) M. C. Puerta and P. Valerga, Coord. Chem. Rev., 1999, 193/195, 977 and refs. therein.
- 2 Y. Yamamoto, X.-H. Han and J.-F. Ma, Angew. Chem., Int. Ed., 2000, 39, 1965; Angew. Chem., 2000, 112, 2041.
- 3 X.-H. Han and Y. Yamamoto, J. Organomet. Chem., 1998, 561, 157.
- 4 Y. Yamamoto, K. Kawasaki and S. Nishimura, J. Organomet. Chem., 1999, 587, 49.
- 5 X.-H. Han and Y. Yamamoto, unpublished work.